

Table 5.2**Important Units of Pressure**

Unit	Relationship or Definition
Pascal (Pa)	$\text{kg}/(\text{m} \cdot \text{s}^2)$
Atmosphere (atm)	$1 \text{ atm} = 1.01325 \times 10^5 \text{ Pa} \approx 100 \text{ kPa}$
mmHg, or torr	$760 \text{ mmHg} = 1 \text{ atm}$